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(21) International Application Number: PCT/EP99/04289 (22) International Filing Date: 21 June 1999 (21.06.99) (30) Priority Data: MI98A001535 3 July 1998 (03.07.98) IT (71) Applicant (for all designated States except US): ZAMBON GROUP S.P.A. [IT/IT]; Via della Chimica, 9, I-36100 Vicenza (IT). (72) Inventors; and (75) Inventors/Applicants (for US only): PAIOCCHI, Maurizio [IT/IT]; Via Valsesia, 86/b, I-20152 Milano (IT). ARRIGHI, Katuscia [IT/IT]; Viale Brianza, 17, I-22060 Cabiato (IT). RUSSO, Laura [IT/IT]; Via Lucio Cornelio Silla, 156/c, I-20153 Milano (IT). VILLA, Marco [IT/IT]; Viale Lunigiana, 10, I-20125 Milano (IT). (74) Agent: LONGONI, Alessandra; Zambon Group S.p.A., Corp. Patent & Trademark Dept., Via Lillo del Duca, 10, I-20091 Bresso (IT).		(81) Designated States: CA, CZ, HU, IL, IN, KR, SI, SK, US, ZA, European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published <i>With international search report.</i>
(54) Title: PROCESS FOR THE PREPARATION OF GABAPENTIN (57) Abstract A process for the preparation of pure gabapentin in anhydrous form by treating an aqueous suspension of gabapentin with 2-methoxyethanol or 2-ethoxyethanol and crystallising with an alcoholic solvent is described.		

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"Process for the preparation of gabapentin"

5 The present invention relates to a process for the preparation of gabapentin and, more particularly, it relates to a process for the preparation of gabapentin in anhydrous form by treating an aqueous suspension of gabapentin with 2-methoxyethanol or 2-ethoxyethanol and crystallising with an alcoholic solvent. Gabapentin, 1-(aminomethyl)cyclohexanacetic acid (The Merck Index, XII ed., page 733, no. 4343), is a known drug with anti-epileptic activity described for the first time
10 by Warner-Lambert Co. in the US patent 4,024,175.

In the literature several processes for the preparation of gabapentin in anhydrous form are reported (see for example the patents US 4,024,175, US 5,068,413 and US 5,091,567). Substantially all these methods foresee a final step for the purification of gabapentin which consists in the treatment of an aqueous solution of a gabapentin
15 salt (generally hydrochloride) through a weak basic ionic exchange resin, the complete evaporation of water from the aqueous gabapentin solution eluted from the resin and the crystallisation from an alcoholic solvent, generally methanol or methanol/isopropanol mixtures.

The complete evaporation of water, described in the above-cited patents, is an
20 expensive and difficult operation with little reproducibility at industrial level, because the product precipitates on the reactor walls and therefore the mixture cannot be stirred. Furthermore, the yield of the subsequent crystallisation depends on the amount of water remained in the product.

In the US patent 4,894,476 is underlined that a further problem related to the known
25 methods for the preparation of gabapentin is the amount of residual solvent contained in the product (column 1, lines 56-59). The same patent describes, as a solution of the problem of the residual solvents, a process for the preparation of anhydrous gabapentin on a large scale that uses monohydrate gabapentin as starting material. However, the preparation of monohydrate gabapentin is
30 particularly troublesome because it foresees, after the usual treatment of an aqueous solution of a gabapentin salt through a weak basic ionic exchange resin, the partial evaporation of water from the aqueous gabapentin solution eluted from the resin, the addition of an alcoholic solvent and the crystallisation.

The resultant pure monohydrate gabapentin is then dissolved in warm methanol,
35 diluted with isopropanol and crystallised obtaining anhydrous gabapentin with a content of methanol and of isopropanol equal to 100 ppm respectively.

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We have now found a method for the preparation of anhydrous gabapentin that allows to obtain the product with a high purity degree and with a low content of residual solvents without using monohydrate gabapentin.

- 5 Therefore, object of the present invention is a process for the preparation of pure gabapentin in anhydrous form comprising (a) the addition of 2-methoxyethanol or of 2-ethoxyethanol to an aqueous gabapentin suspension, (b) the removal of water by azeotropic distillation up to obtain a suspension still containing 20-30% by weight of water with respect to gabapentin, (c) the dilution with an alcoholic solvent and the
- 10 cooling to a temperature from -10°C to +10°C and (d) the filtration and the drying of the crystallised of gabapentin.

The anhydrous gabapentin obtained with the process object of the present invention is characterised by a high purity degree and by a total content of residual solvents lower than 100 ppm.

- 15 Preferably in step (a) of the process 2-methoxyethanol is used.

The amount of 2-methoxyethanol or of 2-ethoxyethanol that is added is not a critical parameter but it will obviously depend on the amount of water present in the aqueous gabapentin suspension to be treated. Generally the aqueous suspension which 2-methoxyethanol or 2-ethoxyethanol is added has a gabapentin

20 concentration from 30% to 40% (w/w). The amount of 2-methoxyethanol or of 2-ethoxyethanol must allow the removal of water by azeotropic distillation up to a residual amount equal to 20%-30% by weight with respect to gabapentin, according to what foreseen in step (b) of the process.

Preferably an amount of 2-methoxyethanol or 2-ethoxyethanol corresponding to 2-5

25 times the amount by weight of gabapentin, still more preferably corresponding to 2-3 times by weight, is used.

In step (b) the gabapentin suspension is heated under reduced pressure by distilling the water/2-methoxyethanol or 2-ethoxyethanol mixture. Preferably a temperature from 40°C to 50°C and a pressure from 50 to 80 mmHg are used.

- 30 In the subsequent step (c), the suspension is diluted with an alcoholic solvent.

The amount of alcoholic solvent is generally from 4 to 10 times by weight with respect to gabapentin, preferably from 4 to 6 times by weight.

Examples of alcoholic solvents are linear or branched C₁-C₄ alcohols such as methanol, ethanol, *n*-propanol, isopropanol, *n*-butanol, isobutanol, *sec*-butanol, *tert*-butanol and mixtures thereof.

35

Isopropanol is preferably used.

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After optional heating to a temperature from 50°C to 70°C, the mixture is cooled to a temperature from -10°C to +10°C, preferably from -5°C to 0°C.

5 The subsequent step (d) foresees the filtration and the drying of the crystallised according to conventional techniques so allowing to obtain gabapentin in pure anhydrous form (HPLC titre > 99.5%) and with a content of residual solvents (alcoholic solvent + 2-methoxyethanol or 2-ethoxyethanol) lower than 100 ppm.

10 The aqueous gabapentin suspension used as starting product in the process object of the present invention is generally an aqueous suspension obtained after elution from a weak basic ionic exchange resin, according to one of the methods described in the literature, and concentration.

15 Preferably the aqueous gabapentin suspension used as starting product comes from the elution of a gabapentin hydrochloride solution through a Relite EXA10 resin. The aqueous solution obtained after crossing through this resin results to be much more concentrated with respect to those obtained after crossing through the resins described in the literature, giving a significant industrial advantage.

20 Therefore a further object of the present invention is a process for the preparation of pure gabapentin in anhydrous form comprising the preparation of an aqueous gabapentin solution by eluting a gabapentin salt solution through a Relite EXA10 resin, the concentration up to obtain an aqueous gabapentin suspension at 30%-40% by weight and the subsequent treatment according to what foreseen in the already reported steps (a), (b), (c) and (d).

Gabapentin hydrochloride can be prepared according to one of the methods described in the literature.

25 The process object of the present invention has the advantage to avoid the complete removal of water and to be reproducible at industrial level because, during the concentration phase, the mass can be always easily stirred. The used amount of solvent is reduced (high productivity) and the product is obtained in anhydrous form with yields higher than 90%, even in the presence of amounts of water higher than 30% w/w with respect to gabapentin.

The content of residual solvents is extremely low, generally lower than 100 ppm, notwithstanding gabapentin monohydrate is not used.

In order to better illustrate the present invention the following examples are now given.

35

Example 1

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Water (130 g), gabapentin hydrochloride (40 g; HPLC titre 94%), prepared according to the process described in US 4,024,175, and charcoal L4S (1.0 g) were charged into a 250 ml reactor at room temperature.

- 5 After keeping under stirring for 15 minutes, the mixture was filtered through a celite panel and the panel was washed with water (5 g). The solution was pumped with a flux of 15 ml/minute in a column containing Relite EXA10 resin (115 ml). At the end of the addition of the solution, the elution of water in the column went on and the solution was collected following the BRIX index.
- 10 The resultant aqueous gabapentin solution (300 g; HPLC titre 10.06%) was concentrated under vacuum (25 mmHg/50°C) up to obtain a stirrable mixture (still containing 46 g of water).
2-Methoxyethanol (80 ml) was added and a water/2-methoxyethanol mixture (50:50 w/w; 76 g) was distilled under vacuum (40-45°C/60 mmHg).
- 15 Isopropanol (180 ml) was added to the suspension and the mixture was heated at 60°C for 30 minutes.
After cooling at 20°C in about 2 hours, the mixture was kept 2 hours at -5°C/-10°C.
The precipitate was filtered and washed with isopropanol (2x10 ml).
After drying under vacuum at 50°C up to constant weight pure gabapentin (28.3 g)
- 20 was obtained.
HPLC titre=99.8%; K.F.=0.05%
Isopropanol < 30 ppm
2-methoxyethanol < 70 ppm.

Example 2

- 25 2-Methoxyethanol (200 ml) and gabapentin hydrochloride (100 g; HPLC titre 92%) were charged into a 0.5 l reactor, equipped with mechanic stirring, condenser and thermometer, kept under inert atmosphere.
The suspension was heated at internal 50-55°C for 30 minutes. The insoluble (about 4 g) was filtered and the solution was concentrated under vacuum (external 50°C;
- 30 30 mmHg) distilling about 115 g of 2-methoxyethanol. Water (250 ml) was added to the residue, kept warm.
The mixture was kept under stirring up to obtain a solution.
The solution, cooled and kept at room temperature, was pumped with a flux of 15 ml/minute into a column containing Relite EXA10 resin (300 ml).
- 35 At the end of the addition of the solution, the elution of water in the column went on and the solution was collected following the BRIX index.

- 5 -

The resultant aqueous gabapentin solution (580 g; HPLC titre 12.7%) was heated at 45°C and then charcoal L4S (5 g) was added.

5 The mixture was kept under stirring for 30 minutes and filtered through a celite panel that was washed with water (15 ml).

The aqueous solution was concentrated under vacuum (25 mmHg/50-55°C) up to obtain a stirrable mixture (weight of the distilled about 380 g). 2-Methoxyethanol (180 ml) was added and the mixture was concentrated at an internal temperature of 40°C at 60 mmHg (distilled about 180 ml of water/2-methoxyethanol=1/1).

10 Isopropanol (400 ml) was added to the suspension and, after heating at 60°C, the mixture was kept at that temperature for 30-60 minutes and then was cooled in about 2 hours at 20°C and then at -5°C.

After 2 hours at -5°C, the precipitate was filtered, washed with isopropanol (2x25 ml) and dried under vacuum at 50°C up to constant weight obtaining pure gabapentin

15 (68 g). HPLC titre=100%; K.F.=0.05%

residual solvent (isopropanol + 2-methoxyethanol) < 100 ppm

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Claims

- 1) A process for the preparation of pure gabapentin in anhydrous form comprising
(a) the addition of 2-methoxyethanol or of 2-ethoxyethanol to an aqueous
gabapentin suspension, (b) the removal of water by azeotropic distillation up to
5 obtain a suspension still containing 20-30% by weight of water with respect to
gabapentin, (c) the dilution with an alcoholic solvent and the cooling to a
temperature from -10°C to +10°C and (d) the filtration and the drying of the
crystallised of gabapentin.
- 10 2) A process according to claim 1 wherein 2-methoxyethanol is used.
- 3) A process according to claim 1 wherein an amount of 2-methoxyethanol or 2-
ethoxyethanol corresponding to 2-5 times the amount by weight of gabapentin is
used.
- 4) A process according to claim 3 wherein the amount corresponds to 2-3 times by
15 weight.
- 5) A process according to claim 1 wherein step (b) is carried out at a temperature
from 40°C to 50°C and at a pressure from 50 to 80 mmHg.
- 6) A process according to claim 1 wherein the alcoholic solvent is a linear or
branched C₁-C₄ alcohol selected among methanol, ethanol, *n*-propanol,
20 isopropanol, *n*-butanol, isobutanol, *sec*-butanol, *tert*-butanol and mixtures
thereof.
- 7) A process according to claim 1 wherein the amount of alcoholic solvent is from 4
to 10 times by weight with respect to gabapentin.
- 8) A process according to claim 7 wherein the amount is from 4 to 6 times by
25 weight.
- 9) A process according to claim 6 wherein the solvent is isopropanol.
- 10) A process according to claim 1 wherein, in step (c), the mixture is cooled at a
temperature from -5°C to 0°C.
- 11) A process according to claim 1 further comprising the preparation of an aqueous
gabapentin solution by elution of a gabapentin salt solution through a Relite
30 EXA10 resin, the concentration up to obtain an aqueous gabapentin suspension
at 30%-40% by weight.

INTERNATIONAL SEARCH REPORT

International Application No

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A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07C227/40 C07C229/28

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 4 894 476 A (DONALD E. BUTLER ET AL.) 16 January 1990 (1990-01-16) cited in the application examples 2,3 ---	1
A	US 4 024 175 A (GERHARD SATZINGER ET AL.) 17 May 1977 (1977-05-17) cited in the application example 1 ---	1
A	US 5 068 413 A (KLAUS STEINER ET AL.) 26 November 1991 (1991-11-26) cited in the application examples 3,6 -----	1

☐ Further documents are listed in the continuation of box C.

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Zervas, B

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Information on patent family members

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Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4894476 A	16-01-1990	AT 101116 T	15-02-1994
		AU 624130 B	04-06-1992
		AU 3267889 A	02-11-1989
		CA 1306755 A	25-08-1992
		DE 68912819 D	17-03-1994
		DE 68912819 T	19-05-1994
		DK 212689 A	03-11-1989
		EP 0340677 A	08-11-1989
		ES 2061774 T	16-12-1994
		FI 892067 A	03-11-1989
		HK 1000798 A	01-05-1998
		IE 62958 B	08-03-1995
		JP 2011546 A	16-01-1990
		JP 2619951 B	11-06-1997
		US 4960931 A	02-10-1990
US 4024175 A	17-05-1977	DE 2460891 A	01-07-1976
		AT 340892 B	10-01-1978
		AT 975075 A	15-05-1977
		AU 8774175 A	23-06-1977
		BE 836835 A	18-06-1976
		CA 1052811 A	17-04-1979
		CH 612665 A	15-08-1979
		CH 612666 A	15-08-1979
		CH 612664 A	15-08-1979
		DE 2543821 A	14-04-1977
		DK 581475 A,B,	22-01-1976
		ES 443723 A	16-04-1977
		FI 753613 A,B,	22-06-1976
		FR 2294697 A	16-07-1976
		GB 1465229 A	23-02-1977
		IE 42382 B	30-07-1980
		JP 941538 C	20-02-1979
		JP 51088940 A	04-08-1976
		JP 53024064 B	18-07-1978
		LU 74058 A	20-07-1976
		NL 7514900 A,B,	23-06-1976
		SE 423385 B	03-05-1982
		SE 7514442 A	22-06-1976
		US 4087544 A	02-05-1978
US 5068413 A	26-11-1991	DE 3928184 A	28-02-1991
		AT 98219 T	15-12-1993
		DE 59003771 D	20-01-1994
		DK 414275 T	14-02-1994
		EP 0414275 A	27-02-1991
		ES 2059938 T	16-11-1994
		FI 103506 B	15-07-1999
		IE 63922 B	14-06-1995
		IL 95480 A	29-06-1995
		JP 2839344 B	16-12-1998
		JP 3090054 A	16-04-1991
		PT 95099 A,B	18-04-1991

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